

FACTORS AFFECTING BIOLEACHING IN THE PROCESSING OF NON-METALLICS

IVETA ŠTYRIAKOVÁ

Department of Biotechnology, Institute of Geotechnics of the Slovak Academy of Sciences, Watsonova 45, Košice, SK-043 53, Slovak Republic (bacil@saske.sk)

Abstract: Biotechnological treatment of non-metallics is based on bacterial leaching of raw material and dissolution of Fe. Bacterial iron dissolution ability is dependent on various physicochemical factors as temperature, acidity of solutions, redox potential, rapidity of water circulation and presence of organic sources. The Fe content in the quartz sands and feldspar samples by the biological leaching decreased as much as 60% and by subsequent using of electromagnetic separation of feldspars, the decrease of Fe content in 74% was achieved. However, the application of magnetic separation of quartz sands after bioleaching resulted in total iron removal of 93% and in such combined way prepared product contained 0.024 % of Fe₂O₃. Achieved results on iron removal point to the fact that combination of leaching and magnetic separation enables to obtain product usable in glass and ceramic industry.

Key words: bioleaching, non-metallics, iron, *Bacillus sp.*

1. Introduction

Non-metallic raw materials such as silicates, carbonates and oxides contain no energy source for the microorganisms to utilize. Such materials may be leached by using heterotrophic bacteria and fungi, which require an organic carbon source as a source of energy and carbon for their growth. Bioleaching of non-metallic minerals may be used for the recovery of metals from low-grade ores and minerals as well as for the beneficiation of mineral raw materials, recovery of metals from sediments, and heavy metal detoxification of soils and solid residues (JAIN, 2004).

Heterotrophic bacteria and fungi have the potential for producing acidic metabolites that are able to solubilize oxide, silicate, carbonate and hydroxide minerals by reduction, acidolysis and complexation mechanisms (JAIN, 2004).

The use of fungi presents a problem, however. The material to be leached is often adsorbed to or enclosed by the fungal mycelium. This is especially undesirable in the beneficiation of coal, bauxite and other materials. In this case the process must either be performed in two separated steps: (1) metabolite production and (2) leaching by the metabolite (GROUDEV and GROUDEVA, 1986; VACHON *et al.*, 1994; BRAND *et al.*, 1999)

Bioleaching processes using heterotrophic bacteria have received little attention although heterotrophs in the genera *Bacillus* and *Pseudomonas* have been found effective in the bioleaching of non-sulfidic minerals (KARAVAIKO *et al.*, 1980). Naturally occurring silicates contain oxidic iron minerals as coatings on grains or impregnations in the matrix. The extent of iron removal from industrial silicate minerals depends on the mineralogy and distribution of iron in silicate rocks. For this reason, bioleaching studies with industrial minerals for their beneficiation have

examined the kinetics of iron dissolution from the siliceous matrix (VEGLIO, 1997). In the present works, iron reduction was monitored as a measure of bacterial activity in the bioleaching of non-metallic raw materials. The aim of this was to remit on bacterial kind of *Bacillus spp.* can be used for bioleaching and the effect of factors on the extent of Fe removal from non-metallics.

2. Materials and methods

The iron – bearing minerals (goethite, mica, smectite) decrease the quality of non-metallic material. Bioleaching medium containing NaH_2PO_4 – 0.5g/l, $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ – 0.5g/l, $(\text{NH}_4)_2 \text{SO}_4$ – 1.0g/l, NaCl – 0.2g/l. However, medium was changed six times during bioleaching at 17 days intervals under aseptic conditions. The carbon sources were glucose, sucrose, galactose, technical-grade sucrose and molasses (all at 20 g/l).

The samples were inoculated with a mixture of both *Bacillus cereus*, *Bacillus mycoides* and *Bacillus pumilus* strains. The flasks were incubated statically for 3 or 4 months at 28°C. The abiotic controls were cultivated under the same conditions. After incubation, the culture solutions were separated from the biomass by means of membrane filtration.

Quantitative changes in the solid phase were measured with a Model 30 Varian atomic absorption spectrometer (Varian, Inc., Melbourne, Vic., Australia). Dissolved Fe^{2+} and Fe^{3+} were measured spectrophotometrically using the phenantroline method (HERRERA *et al.*, 1989; STUCKY and ANDERSON, 1981).

Headspace gas was analyzed for CO_2 concentration using a gas chromatograph Chrom 5 (L.P. Praha, Prague, Czech Republic) equipped with a TC detector.

Bacterial oxidation of sucrose and glucose was analyzed as described by DUBOIS *et al.* (1956).

Morphological and chemical changes in the surfaces of individual mineral grains were investigated by scanning electron microscopy (SEM) and energy-dispersion microanalysis (EDS), respectively. Mineral samples were coated with carbon and examined under a TESLA BS 340 (TESLA Elmi, Brno, Czech Republic) scanning electron microscope.

Solid residues were analyzed by X-ray diffraction using a Philips X'Pert SW–binary diffractometer with $\text{CuK}\alpha$ radiation (40 kV, 50 mA), equipped with an automatic divergence slit, sample spinner, and a graphite secondary monochromator. Data were collected for 2-60 °2 θ with a step width of 0.05° and a counting time of 30 sec per 0.05°.

Dry electromagnetic separation was carried out using a laboratory high gradient magnetic separator with the induction of magnetic field at 1.3 T.

3. Results and discussion

The iron-bearing minerals can be easily removed by magnetic separation, and ultra-fine iron particles are difficult to treat by conventional mineral processing methods, biochemical leaching appears to be the alternative for the effective removal of iron minerals. Biotechnological treatment is based on bacterial leaching of raw

material and dissolution of Fe. These processes are in last decades often observed *in situ* at weathering of silicate minerals in soil. In order to optimize a leaching process, it is necessary to understand the nature of the biotic reactions. Factors that influence these reactions include the kind of microbial population as well as the mineralogical properties of the raw materials to be leached and also physicochemical parameters:

3.1 Microbial population

Heterotrophic bacteria of *Bacillus* strains are ubiquitous in non-metallic deposit (ŠTYRIKOVÁ and ŠTYRIAK, 2002), they are non-pathogenic and are also facultative anaerobic bacteria enabling an easy manipulation during non-metallic treatment, increased rapidly in the numbers of cell during non-metallic treatment and targetly produced organic acid in silicate structures.

The leaching effect, observed by measurement of Fe^{2+} and Fe^{3+} concentration in solution, showed higher activities bacterial kind of *Bacillus* spp. isolated from Bajkal Lake and also using of yeast *Saccharomyces* sp. during bioleaching of quartz sands. However, every kinds of *Bacillus* spp. isolated from Slovak deposit and from Bajkal Lake were very active in iron reduction and dissolution during bioleaching of feldspar raw material.

The results of experiments confirmed the ability of bacteria of *Bacillus* genus to corrode silicate minerals in feldspar raw materials and quartz sands with subsequent extraction of intergranularly bound iron and smectite as unrequested admixture in non-metallic raw materials (ŠTYRIKOVÁ *et al.*, 2003b, ŠTYRIKOVÁ *et al.*, 2007a).

3.2 Mineralogical properties

The effect of biological leaching is dependent from mineralogical composition of certain raw material and the form of iron binding. Bioleaching was effective for the removal of surface layers comprising fine iron minerals, fine-grained mica and iron smectite fraction from non-metallics. Quartz sands containing mineral impurities in form siderite and mica were successfully treated with bioleaching and elutriation. In the process, poorly crystalline Fe-oxides that sealed siderite nodules were released due to the bacterial action from intergranular space of quartz sands. These Fe-oxides formed a fine-grained fraction with Fe-bearing minerals and fine mica fraction which were subsequently removed by elutriation process (ŠTYRIKOVÁ *et al.*, 2003c). Other quartz sands showed the presence of iron smectite coatings in quartz grains. The bioleaching treatment removed these surface coatings, presumably through the action of organic acid produced by the bacteria, and released fine clay particles from quartz grains. Independent iron minerals were removed by magnetic separation (ŠTYRIKOVÁ *et al.*, 2007a).

Magnetic separation removed independent particles of iron bearing minerals such as biotite and ilmenite. The extent of iron removal moved approximately 60 - 74% in combination of bioleaching and magnetic separation of feldspar treatments (ŠTYRIKOVÁ *et al.*, 2006). The application of magnetic separation of quartz sands after bioleaching resulted in total iron removal of 93 % (ŠTYRIKOVÁ *et al.*, 2007a).

2.3 Temperature

At Slovakia, there is often colder weather. That is why active bacteria with cold shock proteins in their genomes should be suitable for the use at leaching of non-metallic raw materials under such climatic conditions. In view of the low temperature of water at the bottom of Baikal Lake, it can be assumed that cold-adapted microorganisms are found in the sediments. Psychrophilic and psychrotrophic microorganisms are able to grow at temperatures around 0 °C. Psychrophilic microorganisms have an optimum growth temperature ≤ 15 °C and do not grow above 20 °C; psychrotrophic (psychrotolerant) microorganisms have their optimum growth above 15 °C (MARGESIN *et al.*, 2003). Mesophilic bacteria prefer temperatures between 20 and 40 °C, but some mesophiles can slowly grow at low temperatures. Exposure of bacterial cells to a cold shock leads to the synthesis of cold-shock proteins (csp) (FRANCIS *et al.*, 1997). Psychrotrophic and mesophilic strains can be differentiated using PCR by the primers BcFF2, BcAPF1, and BcAPR1 for the amplification of mesophilic/psychrotrophic cspF and psychrotrophic cspA genes (FRANCIS *et al.*, 1998). In the case of mesophilic strains, one 284 bp PCR product was amplified, in the case of psychrotrophic strains, two 160 bp and 284 bp PCR products were obtained. Thus, as a second step for the verification of the quality of DNA isolated, the PCR identification of this cold-shock protein was chosen. From the results it follows that the DNA from psychrotrophic microorganisms was amplified. As the psychrotrophic *Bacillus mycoides* was isolated from Baikal Lake sediments, the DNAs from psychrotrophic *Bacillus cereus* CCM145 and mesophilic *Bacillus thuringiensis* CCM19^T were used as the control in PCR, too (ŠPANOVÁ, 2006).

2.4 Eh and pH of solutions

Heterotrophic bacteria formed anaerobic conditions by microbial respiration after each replacement of medium after 1 day. The creation of Eh gradually decreased from -110 mV in solution to -380 mV in sample bed (ŠTYRIAKOVÁ *et al.*, 2006). The bacteria of *Bacillus spp.* were inoculated into media with sample at the beginning of each batch test, it was assumed that gases were produced and pH was lowered from 7 to 4 due to the fermentation of an organic source during 5 days. Cumulative measurement of CO₂ was made only for the first four days of incubation in the glucose-sucrose medium. The results showed the formation of 27 mg CO₂ from 2 g sucrose and 4 mg CO₂ from 2 g glucose. These values are extremely low and may reflect a high production of organic acids (ŠTYRIAKOVÁ *et al.*, 2007b).

2.5 Water circulation

Experimental studies of the effect of microorganisms on iron removal were performed in either closed (batch) or open (percolated columns) reaction vessels. The availability of carbon source and its concentration in the medium is found to be crucial for the growth of the heterotrophic organisms and metabolite production (CASTRO *et*

al., 2000). Our results from chemical analyses of leaches and solid phases showed that iron dissolution from non-metallic raw materials via aqueous – phase transport in the batch experiment was increased by the sufficient consumption of organic source, thereby allowing a higher increase in the extent of Fe removal from natural sample materials. In contrast to the batch bioleaching, a lower Fe dissolution was observed by biological leaching in percolator columns (ŠTYRIAKOVÁ *et al.*, 2006).

3.6 Organic materials

The effect of carbon source was tested in media inoculated with *B. cereus* by the addition of glucose, sucrose, galactose, technical-grade sucrose and molasses. The results showed that the extent of iron dissolution was higher in the presence technical-grade sucrose and molasses after 6 days of bioleaching of non-metallics. Molasses can be used as a relatively inexpensive organic carbon source for the heterotrophs, leading to the formation of metabolites, most notably organic acids acting as leaching agents (ŠTYRIAKOVÁ *et al.*, 2007b).

4. Conclusion

The results were presented in the many publications may serve as a starting point for development of flexible biotechnology for non-metallic's beneficiation, where either bacterial Fe³⁺ dissolution and Fe²⁺ reduction would be managed, depending on the many physicochemical factors.

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